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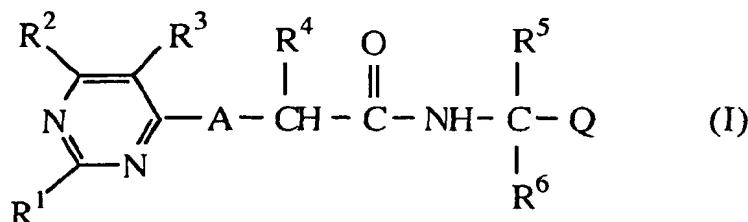
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81925 München (DE)(54) **PYRIMIDINYLOXYALKANOIC AMIDE DERIVATIVES AND FUNGICIDES FOR AGRICULTURAL AND HORTICULTURAL USE**

(57) The present invention provides a pyrimidinyloxyalkanamide derivative represented by Formula (I):



(wherein R¹ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, or the like, R² represents a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, or the like, R³ represents a hydrogen atom, a C₁ - C₆ alkyl group, a halogen atom, or the like, R⁴ represents a hydrogen atom, a C₁ - C₆ alkyl group, or the like, Q represents a cyano group, a C₁ - C₆ alkoxy carbonyl group, or the like, and A represents an oxygen atom or a sulfur atom) and a novel agricultural or horticultural fungicide including the same as an active ingredient.

The agricultural or horticultural fungicides according to the present invention have excellent controlling effects with regard to rice blast, which are apparently excellent controlling effects as compared with the known compounds. In addition, the agricultural or horticultural fungicides of the present invention also exhibit excellent residual activity, as well as persistence after rain without any negative effects on the subject plants.

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Description**Technical Field**

5 [0001] The present invention relates to novel pyrimidinyloxyalkanamide derivatives and agricultural or horticultural fungicides containing the same as active ingredients.

Background Art

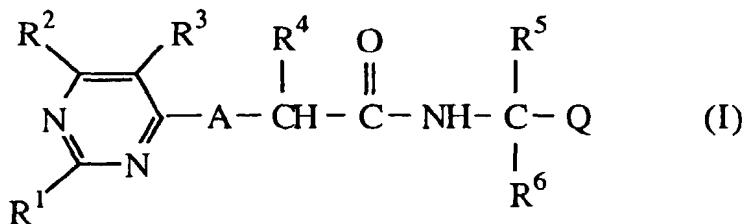
10 [0002] Japanese Patent Application, First Publication, No. Sho 63 - 132867 discloses that aryloxycarboxylic acid derivatives have fungicidal activities. In the previous application, compounds having 2-pyrimidinyl group without any substituents in the aryl group are disclosed; however, they do not exhibit adequate fungicidal activities. In addition, there is no disclosure of compounds having 4-pyrimidinyl group with substituents.

15 [0003] Recently, conventional fungicides may not exhibit adequate fungicidal activities because of the emergence of resistant fungi after repeated use of the fungicides. In addition, as well as because of environmental problems, it is desired to provide novel fungicides which can efficiently control harmful fungi even at low concentrations. The present invention provides novel pyrimidinyloxyalkanamide derivatives exhibiting superior fungicidal activities.

20 [0004] Therefore, the present inventors have synthesized various novel pyrimidinyloxyalkanamide derivatives and have carried out extensive research in connection with their effects on the biological activities of fungi. As a result, the present inventors have found that the novel compounds according to the present invention have wide spectrum of fungicidal activity, and exhibit excellent fungicidal activities with regard to rice blast and the like, while at the same time do not hinder desirable plant growth.

Disclosure of the Invention

25 [0005] The present invention provides pyrimidinyloxyalkanamide derivatives represented by Formula (I):



40 {wherein R¹ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyloxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₆ alkylthio group, a C₁ - C₆ alkylsulfinyl group, a C₁ - C₆ alkylsulfonyl group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a C₃ - C₆ cycloalkylthio group, a C₁ - C₆ alkylamino group, a di(C₁ - C₆ alkyl)amino group, a halogen atom, a phenyl group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom), or a phenoxy group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom),

45 R² represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₂ - C₆ alkenyl group, a C₂ - C₆ alkynyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₂ - C₆ haloalkenyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyloxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₄ haloalkoxy group, a C₁ - C₆ alkylthio group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a C₃ - C₆ cycloalkylthio group, a C₁ - C₆ alkylamino group, a di(C₁ - C₆ alkyl)amino group, a halogen atom, or a phenyl group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom).

50 R³ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a (C₁ - C₆ alkyl)carbonyl group, a (C₁ - C₆ alkoxy)carbonyl group, a halogen atom, a nitro group, or a cyano group, or

55 R² and R³, together with the carbon atom to which they are bonded, form a saturated 6-membered ring or an unsaturated 5-membered or 6-membered ring.

R^4 represents a hydrogen atom, a $C_1 - C_6$ alkyl group, a $C_3 - C_6$ cycloalkyl group, or a $C_1 - C_4$ haloalkyl group,
 R^5 and R^6 represents independently a hydrogen atom, a $C_1 - C_6$ alkyl group, a $C_2 - C_6$ alkenyl group, a $C_3 - C_6$ cycloalkyl group (which may be substituted by a halogen atom or a $C_1 - C_6$ alkyl group), a $C_3 - C_6$ cycloalkyl $C_1 - C_6$ alkyl group, a $C_1 - C_6$ alkoxy $C_1 - C_6$ alkyl group, or a $C_1 - C_4$ haloalkyl group.

R^5 and R^6 , together with the carbon atom to which they are bonded, form a 5-membered ~ 7-membered cycloalkyl group (which may be substituted by a $C_1 - C_6$ alkyl group), or a heterocycl group (which may be substituted by a $C_1 - C_6$ alkyl group),

Q represents a cyano group or a group of a formula: $-COR^7$ [wherein R^7 represents a $C_1 - C_6$ alkyl group, a $C_3 - C_6$ cycloalkyl group (which may be substituted by a halogen atom or a $C_1 - C_6$ alkyl group), a $C_1 - C_4$ haloalkyl group, a $C_1 - C_6$ alkoxy group, a $C_2 - C_6$ alkenyloxy group, a $C_2 - C_6$ alkynyoxy group, or a $C_3 - C_6$ cycloalkyloxy group], and

A represents an oxygen atom or a sulfur atom}, as well as an agricultural or horticultural fungicide including the pyrimidinyloxyalkanamide derivative as an active ingredient.

[0006] The terms employed in the present invention are defined in the following. In the present invention, for example, in the case of the expression " $C_1 - C_6$ ", the group shown after " $C_1 - C_6$ " has 1 to 6 carbon atoms.

[0007] The term " $C_1 - C_6$ alkyl group" is used herein to mean a straight-chain or branched alkyl group including, for example, a methyl group, ethyl group, n -propyl group, isopropyl group, n -butyl group, isobutyl group, *sec*-butyl group, *tert*-butyl group, n -pentyl group, isopentyl group, neopentyl group, n -hexyl group, isohexyl group, 3,3-dimethylbutyl group, or the like.

[0008] As the term " $C_3 - C_6$ cycloalkyl group", there can be mentioned, for example, a cyclopropyl group, cyclopentyl group, cyclohexyl group, or the like.

[0009] As the term " $C_3 - C_6$ cycloalkyl $C_1 - C_6$ alkyl group", there can be mentioned, for example, a cyclopropylmethyl group, cyclopentylmethyl group, cyclohexylmethyl group, or the like.

[0010] The term " $C_1 - C_4$ haloalkyl group" is used herein to mean a halogen-substituted straight-chain or branched alkyl group including, for example, a fluoromethyl group, chloromethyl group, bromomethyl group, difluoromethyl group, dichloromethyl group, dibromomethyl group, trifluoromethyl group, chlorodifluoromethyl group, pentafluoroethyl group, or the like.

[0011] The term " $C_2 - C_6$ alkenyl group" is used herein to mean a straight-chain or branched alkenyl group including, for example, a vinyl group, 1-propenyl group, allyl group, isopropenyl group, 1-butenyl group, 2-butenyl group, or the like.

[0012] The term " $C_2 - C_6$ alkynyl group" is used herein to mean a straight-chain or branched alkynyl group including, for example, an ethynyl group, 1-propynyl group, 2-propynyl group, 1-butynyl group, 2-butynyl group, 3-butynyl group, 4-methyl-1-pentyne group, 3-methyl-1-pentyne group, or the like.

[0013] The term "halogen atom" is used herein to mean a fluorine atom, chlorine atom, bromine atom, or iodine atom.

[0014] The term " $C_1 - C_6$ alkoxy group" is used herein to mean a straight-chain or branched alkoxy group including, for example, a methoxy group, ethoxy group, n -propoxy group, isopropoxy group, n -butoxy group, isobutoxy group, *sec*-butoxy group, *tert*-butoxy group, n -pentyloxy group, isopentyloxy group, n -hexyloxy group, or the like.

[0015] The term " $C_2 - C_6$ alkenyloxy group" is used herein to mean a straight-chain or branched alkenyloxy group including, for example, an allyloxy group, isopropenyoxy group, 2-butenyloxy group, or the like.

[0016] The term " $C_2 - C_6$ alkynyoxy group" is used herein to mean a straight-chain or branched alkynyoxy group including, for example, 2-propynyoxy group, 2-butynyoxy group, 3-butynyoxy group, or the like.

[0017] As the " $C_3 - C_6$ cycloalkyloxy group", there can be mentioned, for example, a cyclopropyloxy group, cyclopentyloxy group, cyclohexyloxy group, or the like.

[0018] The term " $C_1 - C_4$ haloalkoxy group" is used herein to mean a halogen-substituted straight-chain or branched alkoxy group including, for example, a fluoromethoxy group, difluoromethoxy group, trifluoromethoxy group, pentafluoroethoxy group, or the like.

[0019] The term " $C_1 - C_6$ alkylthio group" is used herein to mean a straight-chain or branched alkylthio group including, for example, a methylthio group, ethylthio group, n -propylthio group, isopropylthio group, n -butylthio group, isobutylthio group, *sec*-butylthio group, *tert*-butylthio group, n -hexylthio group, or the like.

[0020] The term " $C_1 - C_6$ alkylsulfinyl group" is used herein to mean a straight-chain or branched alkylsulfinyl group including, for example, a methylsulfinyl group, ethylsulfinyl group, n -propylsulfinyl group, isopropylsulfinyl group, n -butylsulfinyl group, isobutylsulfinyl group, *sec*-butylsulfinyl group, *tert*-butylsulfinyl group, n -hexylsulfinyl group, or the like.

[0021] The term " $C_1 - C_6$ alkylsulfonyl group" is used herein to mean a straight-chain or branched alkylsulfonyl group including, for example, a methylsulfonyl group, ethylsulfonyl group, n -propylsulfonyl group, isopropylsulfonyl group, n -butylsulfonyl group, isobutylsulfonyl group, *sec*-butylsulfonyl group, *tert*-butylsulfonyl group, n -hexylsulfonyl group, or the like.

[0022] The term " $C_1 - C_6$ alkylamino group" is used herein to mean a straight-chain or branched alkylamino group including, for example, a methylamino group, ethylamino group, n -propylamino group, isopropylamino group, n -

butylamino group, isobutylamino group, *sec*-butylamino group, *tert*-butylamino group, *n*-hexylamino group, or the like.

[0023] As the term "di(C₁ - C₆ alkyl)amino group", there can be mentioned, for example, a dimethylamino group, diethylamino group, dipropylamino group, dibutylamino group, or the like.

[0024] The term "(C₁ - C₆ alkyl)carbonyl group" is used herein to mean a straight-chain or branched alkylcarbonyl group including, for example, an acetyl group, propionyl group, butyryl group, isobutyryl group, or the like.

[0025] The term "(C₁ - C₆ alkoxy)carbonyl group" is used herein to mean a straight-chain or branched alkoxy carbonyl group including, for example, a methoxycarbonyl group, ethoxycarbonyl group, *n*-propoxycarbonyl group, isopropoxycarbonyl group, *n*-butoxycarbonyl group, isobutoxycarbonyl group, *sec*-butoxycarbonyl group, *tert*-butoxycarbonyl group, *n*-pentyloxycarbonyl group, *n*-hexyloxycarbonyl group, or the like.

[0026] The term "heterocyclyl group" is used herein to mean a saturated cyclyl group including at least one oxygen atom or sulfur atom as a constituent atom, including, for example, a 3-oxolanyl group, 4-oxanyl group, 3-thiolanyl group, 4-thianyl group, or the like.

[0027] The compounds represented by Formula (I) according to the present invention can exist as optical isomers by virtue of the presence of one or more asymmetric carbon atoms in a molecule. The present invention relates to all such diastereomers, enantiomers, and mixtures thereof.

[0028] The preferred compounds represented by Formula (I) according to the present invention are those in which:

R¹ represents a hydrogen atom, methyl group, cyclopropyl group, methylthio group, ethylthio group, allylthio group, propargylthio group, methoxy group, ethoxy group, phenoxy group, or phenyl group,

R² represents a methyl group, ethyl group, isopropyl group, trifluoromethyl group, chlorodifluoromethyl group, difluoromethyl group, dichloromethyl group, dibromomethyl group, methoxy group, methylthio group, or chlorine atom,

R³ represents a hydrogen atom, methyl group, ethyl group, chlorine atom, or bromine atom,

R⁴ represents a hydrogen atom, methyl group, or ethyl group,

R⁵ represents a hydrogen atom, methyl group, ethyl group, or *n*-propyl group,

R⁶ represents a methyl group, ethyl group, *n*-propyl group, isopropyl group, *n*-butyl group, isobutyl group, *sec*-butyl group, *tert*-butyl group, cyclopropyl group, cyclopentyl group, or dichloromethyl group, and

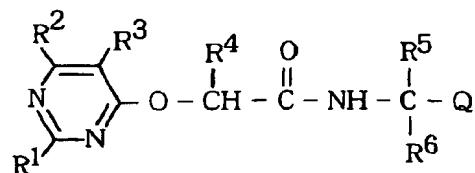
Q represents a cyano group, acetyl group, propionyl group, methoxycarbonyl group, or ethoxycarbonyl group.

[0029] Next, representative examples of the compounds represented by Formula (I) according to the present invention are listed in Tables 1 ~ 15. However, it should be understood that the present invention is not limited to these compounds. Compound Numbers given in the Tables will be referred to in the subsequent description.

[0030] In the tables, "Me" means a methyl group, "Et" means an ethyl group, "*n*-Pr" means an *n*-propyl group, "*i*-Pr" means an isopropyl group, "*n*-Bu" means an *n*-butyl group, "*i*-Bu" means an isobutyl group, "*s*-Bu" means a *sec*-butyl group, "*t*-Bu" means a *tert*-butyl group, and "Ph" means a phenyl group. Therefore, for example "Ph(4-Cl)" means 4-chlorophenyl group.

[0031] "Isomer A" represents an A-configurational diastereomer, "Isomer B" represents a B-configurational diastereomer, and "Isomer M" represents a mixture of an A-configurational diastereomer and a B-configurational diastereomer. "Isomer RA" represents an A-configurational diastereomer wherein the acid moiety is an optically active (R-configurational) one, "Isomer RB" represents a B-configurational diastereomer wherein the acid moiety is an optically active (R-configurational) one, and "Isomer RM" represents a mixture of the diastereomers wherein the acid moiety is an optically active (R-configurational) one. "Isomer SA" represents an A-configurational diastereomer wherein the acid moiety is an optically active (S-configurational) one, "Isomer SB" represents a B-configurational diastereomer wherein the acid moiety is an optically active (S-configurational) one, and "Isomer SM" represents a mixture of the diastereomers wherein the acid moiety is an optically active (S-configurational) one. "A-configurational diastereomer" means a low-polar diastereomer separated by column chromatography on silica gel, high performance liquid chromatography, or the like, while "B-configurational diastereomer" means a high-polar diastereomer separated in the same manner as mentioned above.

Table 1



Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Isomer
A- 1	H	Me	Cl	Me	Me	i-Pr	CN	115~117	M
A- 2	H	Me	Cl	Me	Me	Me	CN		
A- 3	H	Et	CF ₃	H	Me	i-Pr	CN		
A- 4	H	Et	F	Me	Me	i-Pr	CN		
A- 5	H	Et	Cl	Me	Me	i-Pr	CN	129~131	M
A- 6	H	Et	Cl	Me	Et	Et	CN	133~134	
A- 7	H	Et	Cl	H	Me	i-Pr	CN	1. 5166	
A- 8	H	Et	Cl	H	Me	CH ₂ OMe	CN		
A- 9	H	Et	Br	Me	Me	i-Pr	CN		
A-10	H	Et	I	Me	Me	i-Pr	CN		
A-11	H	Et	—	H	Me	i-Pr	CN		
A-12	H	Et	H	Me	Me	t-Bu	CN	140~142	M
A-13	H	Et	Me	Me	Me	i-Pr	CN	148~150	M
A-14	H	H	Cl	Me	Me	i-Pr	CN		
A-15	H	i-Pr	H	Me	Me	i-Pr	CN	89~ 90	M
A-16	H	i-Pr	H	Me	Me	t-Bu	CN	95~ 97	M
A-17	H	i-Pr	Cl	Me	H	i-Pr	CN		
A-18	H	i-Pr	Cl	Me	H	t-Bu	CN	165~168	M
A-19	H	i-Pr	Cl	Me	H	i-Pr	COOMe		
A-20	H	i-Pr	Cl	Me	H	t-Bu	COOMe		
A-21	H	i-Pr	Cl	Me	Me	i-Pr	CN	132~134	M
A-22	H	i-Pr	Cl	Me	Me	t-Bu	CN	87~ 89	M
A-23	H	i-Pr	Cl	Me	s-Bu	Me	CN		
A-24	H	i-Pr	Cl	Me	—	—	CN		
A-25	H	i-Pr	Cl	Me	Me	i-Pr	COOMe	1. 4987	M
A-26	H	i-Pr	Cl	Me	Me	t-Bu	COOMe	1. 5029	M
A-27	H	n-Pr	Cl	Me	Me	i-Pr	CN	108~110	M

Table 2

Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
A-28	H	n-Pr	Cl	Me	Et	Et	CN	95~97	
A-29	H		Cl	H	Me	i-Pr	CN		
A-30	H	CH=CH ₂	Cl	H	Me	i-Pr	CN		
A-31	H	SMe	Cl	Me	Me	i-Pr	CN	150~151	M
A-32	H	OMe	Cl	Me	Me	i-Pr	CN	140~142	M
A-33	H	O	Cl	H	Me	i-Pr	CN		
A-34	H	OCH ₂ CH=CH ₂	Cl	H	Me	i-Pr	CN		
A-35	H	OCH ₂ C≡CH	Cl	H	Me	i-Pr	CN		
A-36	H	Cl	Cl	Me	Me	i-Pr	CN	128~131	M
A-37	H	Cl	Cl	Me	Me	i-Pr	COOMe		
A-38	H	CF ₃	H	Me	Me	i-Pr	CN	137~139	M
A-39	H	CF ₃	Me	Me	Me	i-Pr	CN	149~151	M
A-40	H	CF ₃	Me	Me	Et	Et	CN	149~151	
A-41	H	CF ₃	Me	Me	Me	t-Bu	CN	150~153	A
A-42	H	CF ₃	Me	Me	Me	t-Bu	CN	125~128	B
A-43	H	CF ₃	Me	Me	Me	CHCl ₂	CN	185~188	M
A-44	H	CF ₃	Me	Me	Me		CN	131~134	M
A-45	H	CF ₃	Me	Me	Me		CN		
A-46	H	CF ₃	Me	Me	Me		CN		
A-47	H	CF ₃	Me	Me	Me		CN		
A-48	H	CF ₃	Me	H	Me	i-Pr	COOMe		
A-49	H	CF ₃	Me	Me	Me	i-Pr	COOMe	74~75	M
A-50	H	CF ₃	Me	Me	H	i-Pr	COOMe		
A-51	H	CF ₃	Me	CF ₃	H	i-Pr	COOMe		
A-52	H	CF ₃	Me	Me	H	i-Pr	COOCH ₂ CH=CH ₂		
A-53	H	CF ₃	Me	Me	H	t-Bu	CN		
A-54	H	CF ₃	Et	Me	Me	i-Pr	CN	140~141	A
A-55	H	CF ₃	Et	Me	Me	i-Pr	CN	149~150	B

Table 3

5	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
10	A-56	H	CF ₃	Et	Me	Me	i-Pr			
15	A-57	H	CF ₃	F	H	Me	i-Pr	CN		
20	A-58	H	CF ₃	Cl	H	Me	i-Pr	CN	1.4871	
25	A-59	H	CF ₃	Cl	H	Me	t-Bu	CN	156~157	
30	A-60	H	CF ₃	Cl	H	Et	Et	CN		
35	A-61	H	CF ₃	Cl	Me	Me	Me	CN	148~151	
40	A-62	H	CF ₃	Cl	Me	Me	Et	CN	141~144	M
45	A-63	H	CF ₃	Cl	Me	Me	n-Pr	CN	106~109	M
50	A-64	H	CF ₃	Cl	Me	Me	i-Pr	CN	157~159	A
	A-65	H	CF ₃	Cl	Me	Me	i-Pr	CN	126~129	B
	A-66	H	CF ₃	Cl	Me	Me	i-Pr	CN	149~150	M
	A-67	H	CF ₃	Cl	Me	Me		CN		
	A-68	H	CF ₃	Cl	Me	Me		CN	124~125	A
	A-69	H	CF ₃	Cl	Me	Me		CN	128~130	B
	A-70	H	CF ₃	Cl	Me	Me		CN	114~116	M
	A-71	H	CF ₃	Cl	Me	Me		CN	130~133	A
	A-72	H	CF ₃	Cl	Me	Me		CN	134~137	B
	A-73	H	CF ₃	Cl	Me	Me		CN	107~109	M
	A-74	H	CF ₃	Cl	Me	Me	i-Bu	CN	91~93	M
	A-75	H	CF ₃	Cl	Me	Me	t-Bu	CN	155~157	A
	A-76	H	CF ₃	Cl	Me	Me	t-Bu	CN	129~132	B
	A-77	H	CF ₃	Cl	Me	Me	t-Bu	CN	134~136	M
	A-78	H	CF ₃	Cl	Me	Et	Et	CN	153~154	
	A-79	H	CF ₃	Cl	Me			CN	125~128	
	A-80	H	CF ₃	Cl	Me			CN	172~174	
	A-81	H	CF ₃	Cl	Me			CN	170~172	M
	A-82	H	CF ₃	Cl	Me			CN	141~143	

Table 4

5	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
10	A-83	H	CF ₃	Cl	Me			CN	160~162	
15	A-84	H	CF ₃	Cl	Me			CN		
20	A-85	H	CF ₃	Cl	Me			CN		
25	A-86	H	CF ₃	Cl	Me			COOMe		
30	A-87	H	CF ₃	Cl	Me			COOMe		
35	A-88	H	CF ₃	Cl	Me			CN	214~216	
40	A-89	H	CF ₃	Cl	Me			CN	180~183	M
45	A-90	H	CF ₃	Cl	Me			CN	174~177	M
50	A-91	H	CF ₃	Cl	Me	H	t-Bu	CN	188~190	M
	A-92	H	CF ₃	Cl	Me	H	s-Bu	CN		
	A-93	H	CF ₃	Cl	Me	Me	i-Pr	COOMe	85~ 86	M
	A-94	H	CF ₃	Cl	Me	H	i-Pr	COOMe	106~108	M
	A-95	H	CF ₃	Cl	Me	Me	i-Pr	COOBt	1.4765	M
	A-96	H	CF ₃	Cl	Me	Me	i-Pr	COO-i-Pr	1.4701	M
	A-97	H	CF ₃	Cl	Me	Me	i-Pr	COOCH ₂ C≡CH		
	A-98	H	CF ₃	Cl	Et	Me	i-Pr	CN	154~156	A
	A-99	H	CF ₃	Cl	Et	Me	i-Pr	CN	116~118	B
	A-100	H	CF ₃	Cl	Et	Me	i-Pr	CN	139~142	M
	A-101	H	CF ₃	Cl	Et	Me	t-Bu	CN	106~109	M
	A-102	H	CF ₃	Cl	Et	Me	s-Bu	CN		
	A-103	H	CF ₃	Cl	n-Pr	Me	i-Pr	CN		

Table 5

	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
5	A-104	H	CH(Cl)Me	Cl	H	Me	i-Pr	CN		
10	A-105	H	CH(Cl)Me	Cl	Me	Me	i-Pr	CN		
15	A-106	H	CH(Cl)Me	Cl	Me	Me	t-Bu	CN		
20	A-107	H	CH(Cl)Me	Cl	Me	Me	i-Pr	COOMe		
25	A-108	H	CH(Cl)Me	Cl	Me	Me	t-Bu	COOMe		
30	A-109	H	CF ₃	Br	Me	Me	i-Pr	CN	176~177	A
35	A-110	H	CF ₃	Br	Me	Me	i-Pr	CN	163~164	B
40	A-111	H	CF ₃	Br	Me	Me	i-Pr	CN	153~154	M
45	A-112	H	CF ₃	Br	Me	Me	t-Bu	CN	170~171	A
50	A-113	H	CF ₃	Br	Me	Me	t-Bu	CN	168~170	B
	A-114	H	CF ₃	Br	Me	Me	t-Bu	CN	147~149	M
	A-115	H	CF ₃	Br	Me	H	t-Bu	CN		
	A-116	H	CF ₃	Br	Me	Me	i-Pr	COOMe	110~111	M
	A-117	H	CF ₃	I	Me	Me	i-Pr	CN		
	A-118	H	CF ₃	CN	Me	Me	t-Bu	CN		
	A-119	H	Ph	H	Me	Me	i-Pr	CN		
	A-120	H	NHPr-i	H	Me	Me	i-Pr	CN		
	A-121	H	N(Me) ₂	H	Me	Me	i-Pr	CN		
	A-122	H	NHPr-i	Cl	Me	Me	i-Pr	CN		
	A-123	H	N(Me) ₂	Cl	Me	Me	i-Pr	CN		
	A-124	H	CF ₃	OMe	Me	Me	t-Bu	CN		
	A-125	H	-CH=CH-CH=CH-	Me	Me	i-Pr	CN	139~141	M	
	A-126	H	- (CH ₂) ₄ -	Me	Me	t-Bu	CN			
	A-127	H	- (CH ₂) ₃ -	Me	Me	t-Bu	CN			
	A-128	▷	CF ₃	Cl	Me	Me	i-Pr	CN	131~132	A
	A-129	▷	CF ₃	Cl	Me	Me	i-Pr	CN	128~130	M
	A-130	▷	CF ₃	Cl	Me	Et	i-Pr	CN		
	A-131	Cl	OMe	H	Me	Me	i-Pr	CN		
	A-132	Cl	SMe	H	Me	Me	i-Pr	CN	1.5381	M
	A-133	CF ₃	Me	Cl	Me	Me	i-Pr	CN		
	A-134	CF ₃	Et	Me	Me	Me	i-Pr	CN		

Table 6

Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
A-135	Me	Me	NO ₂	Me	Me	i-Pr	CN		
A-136	Me	Et	i-Pr	Me	Me	i-Pr	CN	140~143	M
A-137	Me	Et	i-Pr	Me	Me	t-Bu	CN	1.5021	M
A-138	Me	C≡CMe	H	Me	Me	t-Bu	CN		
A-139	Me	CH=CHMe	H	Me	Me	i-Pr	CN		
A-140	Me	CH=C(Cl)Me	H	Me	Me	i-Pr	CN		
A-141	Me	CF ₃	H	Me	Me	i-Pr	CN	111~113	A
A-142	Me	CF ₃	H	Me	Me	i-Pr	CN	1.4672	B
A-143	Me	CF ₃	H	Me	Me	i-Pr	CN	1.4710	M
A-144	Me	CF ₃	H	Me	Me	t-Bu	CN	120~122	A
A-145	Me	CF ₃	H	Me	Me	t-Bu	CN	135~137	B
A-146	Me	CF ₃	H	Me	Me	i-Pr	COOMe		
A-147	Me	CF ₃	Cl	H	Me	i-Pr	CN		
A-148	Me	CF ₃	Cl	Me	H	t-Bu	CN		
A-149	Me	CF ₃	Cl	Me	Me	i-Pr	CN	130~132	A
A-150	Me	CF ₃	Cl	Me	Me	i-Pr	CN	154~156	B
A-151	Me	CF ₃	Cl	Me	Me	t-Bu	CN	133~134	A
A-152	Me	CF ₃	Cl	Me	Me	t-Bu	CN	137~139	B
A-153	Me	CF ₃	Cl	Me	Et	Et	CN	135~137	
A-154	Me	CF ₃	Cl	Me	◀	◀	CN		
A-155	Me	CF ₃	Cl	Me	Me	CH ₂ CH=CH ₂	COOEt		
A-156	Me	CF ₃	Cl	Me	◀	Me	CN	189~191	
A-157	Me	CF ₃	Cl	Me	◀	Me	CN	122~125	M
A-158	Me	CF ₃	Cl	Me	Me	i-Pr	COOMe	122~125	M
A-159	Me	CF ₃	Cl	Me	Me	i-Pr	COOEt	1.4778	M
A-160	Me	CF ₃	Cl	Et	Me	i-Pr	CN		
A-161	Me	CF ₃	Cl	◀	Me	i-Pr	CN		
A-162	Me	Ph	H	Me	Me	i-Pr	CN		
A-163	Me	Ph(4-Cl)	H	Me	Me	i-Pr	CN		

Table 7

5	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
10	A-164	Me	Ph(4-Me)	H	Me	Me	i-Pr	CN		
	A-165	Me	Ph(4-OMe)	H	Me	Me	i-Pr	CN		
15	A-166	Me	CF ₃	Me	Me	Me	i-Pr	CN	105~108	A
	A-167	Me	CF ₃	Me	Me	Me	i-Pr	CN	172~175	B
20	A-168	Me	CF ₃	Me	Me	Me	t-Bu	CN	128~129	A
	A-169	Me	CF ₃	Me	Me	Me	t-Bu	CN	131~132	B
25	A-170	Me	CF ₃	Me	H	Me	i-Pr	CN		
	A-171	Me	CF ₃	Me	H	Me	t-Bu	CN		
30	A-172	i-Pr	CF ₃	H	Me	Me	i-Pr	CN	131~134	M
	A-173	t-Bu	CF ₃	H	Me	Me	i-Pr	CN		
35	A-174	OCH ₂ C≡CH	Me	Cl	Me	Me	i-Pr	CN		
	A-175	OCH ₂ CH=CH ₂	Me	Cl	Me	Me	i-Pr	CN		
40	A-176	OMe	CF ₃	Me	Me	Me	i-Pr	CN	not determined	M
	A-177	OMe	SMe	H	Me	Me	i-Pr	CN	not determined	M
45	A-178	O- 	Me	Cl	Me	Me	i-Pr	CN		
	A-179	NH-i-Pr	CF ₃	H	Me	Me	i-Pr	CN		
50	A-180	N(Me) ₂	CF ₃	H	Me	Me	i-Pr	CN		
	A-181	Ph	Me	COMe	Me	Me	i-Pr	CN		
	A-182	Ph	Me	COOMe	Me	Me	i-Pr	CN		
	A-183	Ph	CF ₃	H	Me	Me	i-Pr	CN	152~154	M
	A-184	Ph	CF ₃	H	Me	Me	t-Bu	CN	155~157	M
	A-185	Ph	CF ₃	H	Me	Me	i-Pr	COOMe		
	A-186	Ph(2-Cl)	CF ₃	H	Me	Me	i-Pr	CN		
	A-187	Ph(3-CN)	CF ₃	H	Me	Me	i-Pr	CN		
	A-188	Ph(3-Me)	CF ₃	H	Me	Me	i-Pr	CN		
	A-189	Ph(3-NO ₂)	CF ₃	H	Me	Me	i-Pr	CN		
	A-190	Ph(4-CF ₃)	CF ₃	H	Me	Me	i-Pr	CN		
	A-191	Ph(4-OMe)	CF ₃	H	Me	Me	i-Pr	CN		
	A-192	SMe	Me	H	Me	Me	i-Pr	CN	106~108	M
	A-193	SMe	Et	H	Me	Me	i-Pr	CN	106~108	M
	A-194	SMe	Et	H	Me	Me	t-Bu	CN	108~110	M

Table 8

	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (n _D ²⁰)	Iso-mer
10	A-195	SMe	Et	Me	Me	Me	i-Pr	CN	105~107	M
15	A-196	SMe	Et	Me	Me	Me	t-Bu	CN	1.5303	M
20	A-197	SMe	Et	Me	Me	Me	i-Pr	COMe		
25	A-198	SMe	Et	Me	Me	Me	i-Pr	COOMe		
30	A-199	SMe	Et	Et	Me	Me	i-Pr	CN	136~138	M
35	A-200	SMe	Et	Et	Me	Me	t-Bu	CN	77~78	M
40	A-201	SMe	Et	i-Pr	Me	Me	i-Pr	CN	134~136	M
45	A-202	SMe	Et	i-Pr	Me	Me	t-Bu	CN	84~85	M
50	A-203	SMe	i-Pr	H	Me	Me	i-Pr	CN	99~102	M
	A-204	SMe	i-Pr	H	Me	Me	i-Pr	COOMe	57~60	M
	A-205	SMe	i-Pr	H	Me	Me	i-Pr	COOMe	75~77	B
	A-206	SMe	i-Pr	H	Me		Me	CN		
	A-207	SMe	i-Pr	H	Me			CN		
	A-208	SMe	i-Pr	Cl	Me	Me	i-Pr	CN	120~122	M
	A-209	SMe	i-Pr	Cl	Me	Me	t-Bu	CN		
	A-210	SMe	i-Pr	Cl	Me	Me	i-Pr	COOMe	41~43	M
	A-211	SMe	i-Pr	Cl	Me	Et	i-Pr	CN		
	A-212	SMe	i-Pr	Cl	Me	Et	Et	CN		
	A-213	SMe	i-Pr	Cl	Me			CN		
	A-214	SMe	n-Pr	H	Me	Me	i-Pr	CN	130~132	M
	A-215	SMe	n-Pr	H	Me	Me	t-Bu	CN	95~97	M
	A-216	SMe	C(Me) ₂ Cl	Cl	Me	Me	i-Pr	CN	132~135	M
	A-217	SMe	C(Me) ₂ Cl	Cl	Me	Me	t-Bu	CN		
	A-218	SMe	C(Me) ₂ Cl	Cl	Me	Me	i-Pr	COOMe		
	A-219	SMe	CF ₃	H	Me	Me	i-Pr	CN	110~113	M
	A-220	SMe	CF ₃	H	Me	Me	t-Bu	CN	108~110	M
	A-221	SMe	CF ₃	H	Me	Me		CN		
	A-222	SMe	CF ₃	H	H	Me	t-Bu	CN		
	A-223	SMe	CF ₃	Cl	Me	Me	i-Pr	CN	114~116	M
	A-224	SMe	CF ₃	Cl	Me	Me	t-Bu	CN	131~133	M
	A-225	SMe	CF ₃	Cl	Me	Me	i-Pr	COOMe	1.5093	M

Table 9

5	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso- mer
10	A-226	SMe	CF ₃	Cl	Me	Me	t-Bu	COOMe		
15	A-227	SMe	CF ₃	Me	Me	Me	i-Pr	CN	110~112	M
20	A-228	SMe	CF ₃	Me	Me	Me	t-Bu	CN	78~80	M
25	A-229	SMe	CF ₃	Me	Me	Me	i-Pr	COOMe	not determined	M
30	A-230	SMe	Cl	H	Me	Me	t-Bu	CN		
35	A-231	SMe	OMe	H	Me	Me	t-Bu	CN		
40	A-232	SMe	OCHF ₂	H	Me	Me	t-Bu	CN		
45	A-233	SMe	SCH ₂ C≡CH	H	Me	Me	t-Bu	CN		
50	A-234	SMe	SCH ₂ CH=CH ₂	H	Me	Me	t-Bu	CN		
	A-235	SMe	SMe	H	Me	Me	t-Bu	CN		
	A-236	SO ₂ Me	CF ₃	Me	Me	Me	i-Pr	CN	68~70	M
	A-237	SEt	Et	Et	Me	Me	i-Pr	CN	97~99	M
	A-238	SEt	Et	Et	Me	Me	t-Bu	CN	64~66	M
	A-239	SEt	Et	i-Pr	Me	Me	i-Pr	CN	118~119	M
	A-240	SEt	Et	i-Pr	Me	Me	t-Bu	CN	107~108	M
	A-241	SEt	CF ₃	H	Me	Me	i-Pr	CN	104~105	M
	A-242	SEt	CF ₃	H	Me	Me	t-Bu	CN	88~90	M
	A-243	SEt	CF ₃	H	Me	Me	i-Pr	COOMe		
	A-244	SEt	CF ₃	H	Me	Me	t-Bu	COOMe		
	A-245	S- 	Me	Cl	Me	Me	i-Pr	CN		
	A-246	S-i-Pr	Et	Et	Me	Me	i-Pr	CN	72~74	M
	A-247	S-i-Pr	Et	i-Pr	Me	Me	i-Pr	CN	112~114	M
	A-248	S-i-Pr	Et	i-Pr	Me	Me	t-Bu	CN	99~100	M
	A-249	S-i-Pr	CF ₃	H	Me	Me	i-Pr	CN	69~71	M
	A-250	S-i-Pr	CF ₃	H	Me	Me	t-Bu	CN	108~110	M
	A-251	SCH ₂ CH=CH ₂	CF ₃	H	Me	Me	i-Pr	CN	96~98	M
	A-252	SCH ₂ CH=CH ₂	CF ₃	H	Me	Me	t-Bu	CN	96~98	M
	A-253	SCH ₂ C≡CH	CF ₃	H	Me	Me	i-Pr	CN	120~122	M
	A-254	SCH ₂ C≡CH	CF ₃	H	Me	Me	t-Bu	CN	107~109	M

Table 10

Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso-mer
A-255	H	CF ₂ Cl	Br	Me	Me	i-Pr	CN	162~164	M
A-256	H	CF ₂ Cl	Br	Me	Me	t-Bu	CN	141~142	A
A-257	H	CF ₂ Cl	Br	Me	Me	t-Bu	CN	140~142	B
A-258	H	CF ₂ Cl	Br	Me	Me	t-Bu	CN	121~124	M
A-259	H	CF ₂ Cl	Br	Me	Me	i-Pr	COMe	156~159	M
A-260	H	CF ₂ Cl	Br	Me	Et	Et	COMe	177~180	
A-261	H	CF ₂ Cl	Cl	Me	Me	i-Pr	CN	165~166	A
A-262	H	CF ₂ Cl	Cl	Me	Me	i-Pr	CN	165~167	B
A-263	H	CF ₂ Cl	Cl	Me	Me	i-Pr	CN	158~160	M
A-264	H	CF ₂ Cl	Cl	Me	Me	t-Bu	CN	134~136	A
A-265	H	CF ₂ Cl	Cl	Me	Me	t-Bu	CN	118~120	B
A-266	H	CF ₂ Cl	Cl	Me	Me	t-Bu	CN	111~114	M
A-267	H	CF ₃	Br	Me	Me	i-Pr	CN	148~151	RA
A-268	H	CF ₃	Br	Me	Me	i-Pr	CN	125~126	RB
A-269	H	CF ₃	Br	Me	Me	i-Pr	CN	123~125	RM
A-270	H	CF ₃	Br	Me	Me	i-Pr	COMe	143~144	M
A-271	H	CF ₃	Br	Me	Me	i-Pr	COEt		
A-272	H	CF ₃	Br	Me	Me	i-Pr	CO $\text{--} \text{C}_6\text{H}_4\text{--}$		
A-273	H	CF ₃	Br	Me	Me	i-Pr	COCF ₃		
A-274	H	CF ₃	Cl	Me	Et	i-Pr	CN	138~141	M
A-275	H	CF ₃	Cl	Me	Me	$\text{--} \text{C}_6\text{H}_4\text{--} \text{Me}$	CN	113~114	M
A-276	H	CF ₃	Cl	Me	Me	t-Bu	CN	117~118	RA
A-277	H	CF ₃	Cl	Me	Me	t-Bu	CN	120~123	RB
A-278	H	CF ₃	Cl	Me	Me	t-Bu	CN	115~116	SA
A-279	H	CF ₃	Cl	Me	Me	t-Bu	CN	118~121	SB
A-280	H	CF ₃	Cl	Et	Me	t-Bu	CN	123~124	A
A-281	H	CF ₃	Cl	Et	Me	t-Bu	CN	106~109	M
A-282	H	CF ₃	Cl	Me	Me	i-Pr	CN	155~158	RA
A-283	H	CF ₃	Cl	Me	Me	i-Pr	CN	84~85	RB
A-284	H	CF ₃	Cl	Me	Me	i-Pr	CN	111~114	RM
A-285	H	CF ₃	Cl	Me	Me	i-Pr	CN	150~153	SA

Table 11

	Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (n _D ²⁰)	Isomer
5	A-286	H	CF ₃	Cl	Me	Me	i-Pr	CN	94~96	SB
10	A-287	H	CF ₃	Cl	Me	Me	i-Pr	CN	104~106	SM
15	A-288	H	CF ₃	Cl	Me	Me	i-Pr	COMe	151~154	M
20	A-289	H	CF ₃	Cl	Me	Me	i-Pr	COPr-n		
25	A-290	H	CF ₃	Cl	Me	Me	i-Pr	COCH ₂ Cl		
30	A-291	H	CF ₃	Cl	Me	Et	Et	COMe	170~172	
35	A-292	H	CF ₃	Cl	Me	Me	Et	COMe	178~181	M
40	A-293	H	CF ₃	Cl	Me	Me	Me	COMe	201~203	
45	A-294	H	CF ₃	Me	Me	Me	t-Bu	CN	108~110	RA
50	A-295	H	CF ₃	Me	Me	Me	t-Bu	CN	109~112	RB
	A-296	H	CF ₃	Me	Me	Me	i-Pr	COMe	120~123	M
	A-297	H	CF ₃	Me	Me	Me	Et	COMe	155~158	M
	A-298	H	CF ₃	Me	Me	Et	Et	COMe	141~144	
	A-299	Me	CF ₃	Me	Me	Me	i-Pr	COMe	114~117	M
	A-300	Me	CF ₃	Me	Me	Me	Et	COMe	138~141	M
	A-301	Me	CF ₃	Me	Me	Et	Et	COMe	95~98	
	A-302	H	CHBr ₂	Cl	Me	Me	i-Pr	CN	54~57	M
	A-303	H	CHBr ₂	Cl	Me	Me	t-Bu	CN	155~158	M
	A-304	H	CHCl ₂	Cl	Me	Me	i-Pr	CN	160~162	M
	A-305	H	CHCl ₂	Cl	Me	Me	t-Bu	CN	152~154	M
	A-306	H	CHF ₂	Br	Me	Me	i-Pr	CN	166~167	A
	A-307	H	CHF ₂	Br	Me	Me	i-Pr	CN	147~148	B
	A-308	H	CHF ₂	Br	Me	Me	i-Pr	CN	154~155	M
	A-309	H	CHF ₂	Br	Me	Me	t-Bu	CN	138~141	A
	A-310	H	CHF ₂	Br	Me	Me	t-Bu	CN	128~129	B
	A-311	H	CHF ₂	Br	Me	Me	t-Bu	CN	111~113	M
	A-312	H	CHF ₂	Cl	Me	Me	i-Pr	CN	171~172	A
	A-313	H	CHF ₂	Cl	Me	Me	i-Pr	CN	141~144	B
	A-314	H	CHF ₂	Cl	Me	Me	i-Pr	CN	152~155	M
	A-315	H	CHF ₂	Cl	Me	Me	t-Bu	CN	125~127	A
	A-316	H	CHF ₂	Cl	Me	Me	t-Bu	CN	119~122	B

Table 12

Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Iso-mer
A-317	H	CHF ₂	Cl	Me	Me	t-Bu	CN	94~97	M
A-318	H	CHF ₂	Cl	Me	Me	i-Pr	COMe	159~162	M
A-319	H	Ph	Cl	Me	Me	i-Pr	CN	40~42	M
A-320	H	Ph	Cl	Me	Me	t-Bu	CN	1.5401	M
A-321	H	t-Bu	Br	Me	Me	i-Pr	CN	151~153	M
A-322	H	t-Bu	Br	Me	Me	t-Bu	CN	126~129	M
A-323	Me	CF ₃	Br	Me	Me	i-Pr	CN	147~149	A
A-324	Me	CF ₃	Br	Me	Me	i-Pr	CN	141~143	B
A-325	Me	CF ₃	Br	Me	Me	i-Pr	CN	136~137	M
A-326	Me	CF ₃	Br	Me	Me	t-Bu	CN	124~125	A
A-327	Me	CF ₃	Br	Me	Me	t-Bu	CN	142~145	B
A-328	Me	CF ₃	Br	Me	Me	t-Bu	CN	142~144	M
A-329	Me	CF ₃	Br	Me	Me	t-Bu	COMe		
A-330	Me	CF ₃	Br	Me	Me	i-Pr	COMe	122~124	M
A-331	Me	CF ₃	Br	Me	Me	i-Pr	CO ---		
A-332	Me	CF ₃	Br	Me	Me	i-Pr	COEt		
A-333	Me	CF ₃	Br	Me	Et	Et	COMe	132~134	
A-334	Me	CF ₃	Cl	Me	Me	i-Pr	COMe	124~126	M
A-335	Me	CF ₃	Cl	Me	Me	Et	COMe	131~134	M
A-336	Et	CF ₃	Br	Me	Me	i-Pr	CN	105~108	M
A-337	Et	CF ₃	H	Me	Me	i-Pr	CN	90~93	M
A-338	i-Pr	CF ₃	Br	Me	Me	i-Pr	CN	131~133	M
A-339	i-Pr	CF ₃	Br	Me	Me	t-Bu	CN	150~151	M
A-340	i-Pr	CF ₃	H	Me	Me	i-Pr	CN	131~134	M
A-341	i-Pr	CF ₃	H	Me	Me	t-Bu	CN	129~132	M
A-342	OMe	CF ₃	Br	Me	Me	i-Pr	CN	126~129	M
A-343	OMe	CF ₃	Br	Me	Me	t-Bu	CN	118~119	A
A-344	OMe	CF ₃	Br	Me	Me	t-Bu	CN	108~110	B
A-345	OMe	CF ₃	Br	Me	Me	t-Bu	CN	112~115	M
A-346	OEt	CF ₃	Br	Me	Me	i-Pr	CN	111~113	M
A-347	OEt	CF ₃	Br	Me	Me	t-Bu	CN	97~98	M

Table 13

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Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (n _D ²⁰)	Isomer
A-348	OPh	CF ₃	Br	Me	Me	t-Bu	CN	193~194	M
A-349	OPh	CF ₃	Cl	Me	Me	i-Pr	CN	162~165	M
A-350	OPh	CF ₃	H	Me	Me	i-Pr	CN	155~158	M
A-351	OPh(2-Cl)	CF ₃	H	Me	Me	i-Pr	CN		
A-352	OPh(3-Me)	CF ₃	H	Me	Me	i-Pr	CN		
A-353	OPh(4-OMe)	CF ₃	H	Me	Me	i-Pr	CN		
A-354	OPh	CF ₃	Br	Me	Me	i-Pr	CN	170~173	M
A-355	Ph	CF ₃	Cl	Me	Me	i-Pr	CN	152~155	M
A-356	Ph	CF ₃	Cl	Me	Me	t-Bu	CN	166~169	M
A-357	OPh(2-CN)	CF ₃	H	Me	Me	i-Pr	CN		
A-358	OPh(3-NO ₂)	CF ₃	H	Me	Me	i-Pr	CN		
A-359	OPh(4-CF ₃)	CF ₃	H	Me	Me	i-Pr	CN		
A-360	SMe	CF ₃	Br	Me	Me	i-Pr	CN	129~130	M
A-361	SMe	CF ₃	Br	Me	Me	t-Bu	CN	139~142	M
A-362	SO ₂ Me	CF ₃	Br	Me	Me	i-Pr	CN	75~78	M
A-363	SMe	CF ₃	Br	Me	Me	Me	CO^F		
A-364	SMe	CF ₃	Br	Me	Me	Me	CO^{Me}		

Table 14

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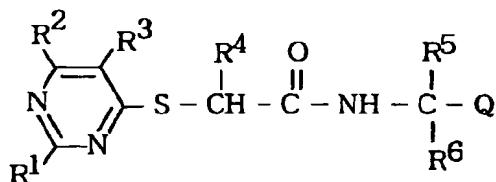
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Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (nD ²⁰)	Isomer
B- 1	H	CF ₃	Br	Me	Me	i-Pr	CN	139~141	M
B- 2	H	CF ₃	Br	Me	Me	t-Bu	CN	158~159	A
B- 3	H	CF ₃	Br	Me	Me	t-Bu	CN	146~147	B
B- 4	H	CF ₃	Br	Me	Me	t-Bu	COOMe		
B- 5	H	CF ₃	Cl	Me	Me	i-Pr	CN	1.5065	A
B- 6	H	CF ₃	Cl	Me	Me	i-Pr	CN	1.5107	B
B- 7	H	CF ₃	Cl	Me	Me	i-Pr	CN	1.5072	M
B- 8	H	CF ₃	Cl	Me	Et	Et	CN		
B- 9	H	CF ₃	Cl	Me	Me	i-Pr	COOMe		
B-10	H	CF ₃	Me	Me	Me	i-Pr	CN	115~116	A
B-11	H	CF ₃	Me	Me	Me	i-Pr	CN	110~112	B
B-12	H	CF ₃	Me	Me	Et	Et	CN		
B-13	H	CF ₃	Me	Me	Me	CHCl ₂	CN		
B-14	H	CF ₃	Me	Me	Me	t-Bu	CN	143~145	A
B-15	H	CF ₃	Me	Me	Me	t-Bu	CN	126~128	B
B-16	H	CF ₃	Me	Me	Me	▷	CN		
B-17	H	i-Pr	Cl	Me	Me	i-Pr	CN	73~75	M
B-18	H	i-Pr	Cl	Me	Me	t-Bu	CN	1.5339	M
B-19	H	i-Pr	Cl	Me	H	i-Pr	CN		
B-20	H	i-Pr	Cl	Me	H	t-Bu	CN		
B-21	H	i-Pr	H	Me	Me	i-Pr	CN	1.5305	M
B-22	H	i-Pr	H	Me	Me	t-Bu	CN	1.5305	M

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Table 15

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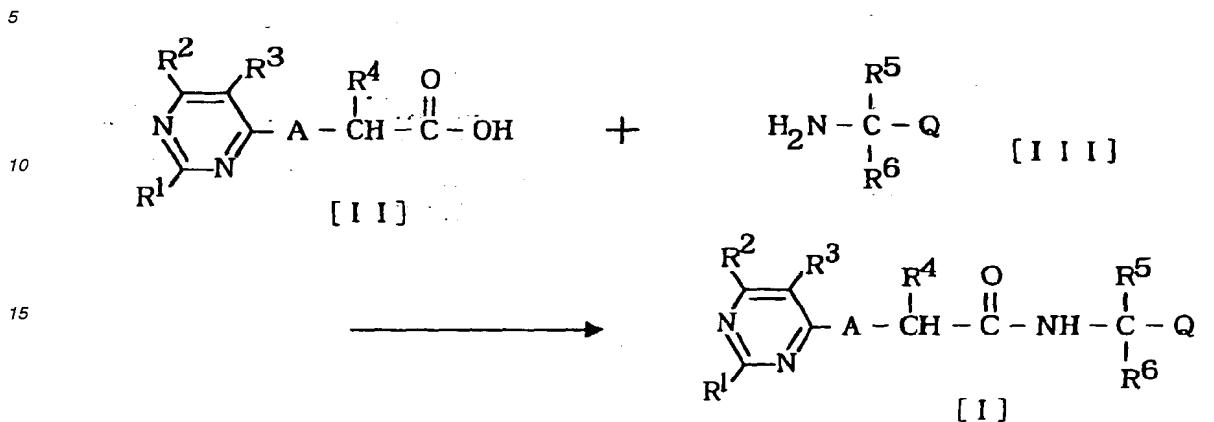
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Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Q	Melting Point (°C) or Refractive Index (n _D ²⁰)	Iso-mer
B-23	Me	CF ₃	H	Me	Me	i-Pr	CN		
B-24	Me	CF ₃	H	Me	Me	t-Bu	CN		
B-25	△	CF ₃	Cl	Me	Me	i-Pr	CN	not determined	A
B-26	△	CF ₃	Cl	Me	Me	i-Pr	CN	117~119	B
B-27	△	CF ₃	Cl	Me	H	i-Pr	COOMe		
B-28	SMe	CF ₃	Cl	Me	Me	i-Pr	CN	102~104	M
B-29	SMe	CF ₃	Cl	Me	Me	t-Bu	CN	132~134	M
B-30	SMe	Ph	H	Me	Me	i-Pr	CN		
B-31	SO ₂ Me	Ph	H	Me	Me	i-Pr	CN		
B-32	SO ₂ Me	Ph	H	Me	Me	i-Pr	CN		
B-33	Me	Ph (2-CN)	H	Me	Me	i-Pr	CN		
B-34	Me	Ph (3-CF ₃)	H	Me	Me	i-Pr	CN		
B-35	Me	Ph (4-NO ₂)	H	Me	Me	i-Pr	CN		

Preparation Process 1

[0033]



(wherein R¹, R², R³, R⁴, R⁵, R⁶, Q, and A have the same meanings as defined above).

25 [0034] The compounds of Formula (I) according to the present invention may be prepared by the reaction of pyrimidinyloxyalkanoic acid derivatives represented by Formula (II) with amines represented by Formula (III) using a condensing agent, in the presence of a catalyst and/or a base, if necessary. The present reaction is generally carried out in a solvent: this solvent can be any solvent that does not hinder the reaction, for example, hydrocarbons such as pentane, hexane, heptane, cyclohexane, petroleum ether, ligroin, benzene, toluene, xylene and the like, halogenated hydrocarbons such as dichloromethane, dichloroethane, chloroform, carbon tetrachloride, chlorobenzene, dichlorobenzene and the like, ethers such as diethyl ether, diisopropyl ether, ethylene glycol dimethyl ether, tetrahydrofuran, dioxane and the like, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone, methyl isobutyl ketone and the like, esters such as methyl acetate, ethyl acetate and the like, nitriles such as acetonitrile, propionitrile, and the like, aprotic polar solvents such as dimethylsulfoxide, N,N-dimethylformamide, sulfolane, and the like, and mixtures of solvents combining solvents selected from the aforementioned.

30 [0035] As the condensing agent, there can be mentioned 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride, N,N'-dicyclohexylcarbodiimide, carbonyldiimidazol, 2-chloro-1,3-dimethylimidazolium chloride, or the like.

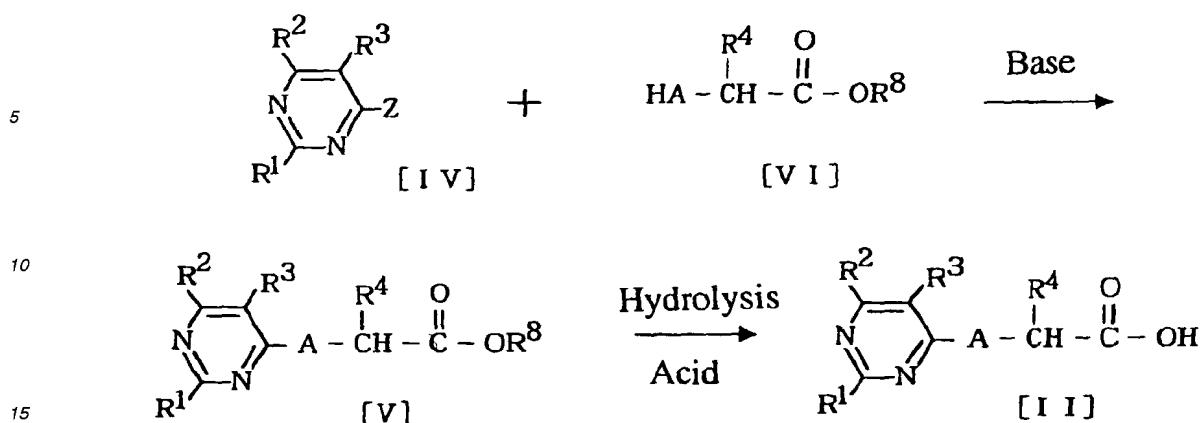
35 [0036] As the catalyst, there can be mentioned, for example, 4-dimethylaminopyridine, 1-hydroxybenzotriazole, dimethylformamide or the like.

40 [0037] The base can be any type of base generally used in this type of reaction. For example, there can be mentioned hydroxides of alkaline metals such as sodium hydroxide, potassium hydroxide and the like, hydroxides of alkaline earth metals such as calcium hydroxide and the like, carbonates of alkaline metals such as sodium carbonate, potassium carbonate and the like, organic bases such as triethylamine, trimethylamine, N,N-dimethylaniline, pyridine, N-methylpiperidine, 1,5-diazabicyclo[4.3.0]non-5-ene (DBN), 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), and the like, and preferably 45 tertiary amines such as triethylamine, pyridine, N-methylpiperidine and the like.

45 [0038] The present reaction is carried out at a temperature in a range from -50°C to 150°C, and preferably in a range of from 0°C to 60°C. The reaction time is preferably in a range from 1 to 30 hours.

50 [0039] Next, the synthesis process for each starting material will be explained.

[0040] The compounds represented by Formula (II) can be synthesized according to, for example, the preparation processes shown below.



(wherein R¹, R², R³, R⁴, and A have the same meanings as defined above, R⁸ represents C₁ to C₆ alkyl group, and Z represents a leaving group such as a halogen atom, or the like).

[0041] The pyrimidinyloxyalkanoic acid derivatives represented by Formula (II) can be prepared, for example, by reacting pyrimidine derivatives represented by Formula (IV) with ester derivatives of alcanoic acids represented by Formula (VI) in the presence of a base to produce ester derivatives of pyrimidinyloxyalkanoic acids represented by Formula (V), and subsequently hydrolyzing the ester derivatives of pyrimidinyloxyalkanoic acids.

[0042] In the reaction schemes described above, the reaction of pyrimidine derivatives represented by Formula (IV) with ester derivatives represented by Formula (VI) is generally carried out in a solvent: this solvent can be any solvent that does not hinder the reaction, for example, hydrocarbons such as pentane, hexane, heptane, cyclohexane, petroleum ether, ligroin, benzene, toluene, xylene and the like, halogenated hydrocarbons such as dichloromethane, dichloroethane, chloroform, carbon tetrachloride, chlorobenzene, dichlorobenzene and the like, ethers such as diethyl ether, diisopropyl ether, ethylene glycol dimethyl ether, tetrahydrofuran, dioxane and the like, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone, methyl isobutyl ketone and the like, acetates such as methyl acetate, ethyl acetate and the like, nitriles such as acetonitrile, propionitrile, and the like, aprotic polar solvents such as dimethylsulfoxide, N,N-dimethylformamide, sulfolane, and the like, and mixture of solvents combining solvents selected from the aforementioned.

[0043] The base can be any type of base generally used in this type of reaction. For example, there can be mentioned inorganic bases such as sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, sodium bicarbonate, sodium hydride, potassium hydride, and the like, or organic bases such as triethylamine, trimethylamine, N,N-dimethylaniline, pyridine, and the like.

[0044] The present reaction is carried out at a temperature in a range from -50°C to 150°C, and preferably in a range of from 0°C to 60°C. The reaction time is preferably in a range from 1 to 30 hours.

[0045] The reaction for obtaining pyrimidinyloxyalkanoic acid derivatives represented by Formula (II) by hydrolysis of ester derivatives of pyrimidinyloxyalkanoic acids represented by Formula (V) is generally carried out in a solvent: this solvent can be any solvent that does not hinder the reaction, for example, water, alcohols such as methanol, ethanol, 2-propanol, and the like, ethers such as diethyl ether, diisopropyl ether, ethylene glycol dimethyl ether, tetrahydrofuran, dioxane and the like, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone, methyl isobutyl ketone and the like, and mixtures of solvents combining solvents selected from the aforementioned.

[0046] The base can be any type of base generally used in this type of reaction. For example, there can be mentioned inorganic bases such as sodium hydroxide, potassium hydroxide and the like.

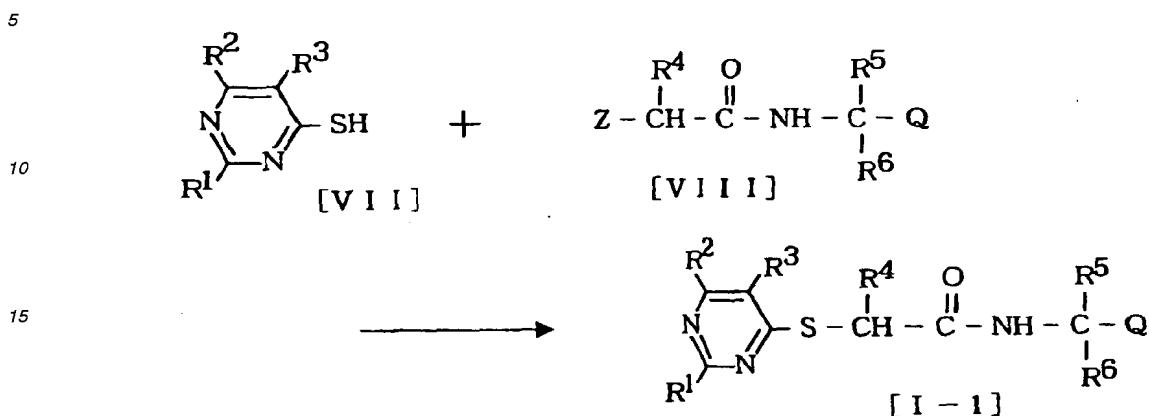
[0047] The present reaction is carried out at a temperature in a range from -50°C to 150°C, and preferably in a range of from 0°C to 60°C. The reaction time is preferably in a range from 1 to 30 hours.

[0048] The compounds represented by Formula (IV) can be synthesized, for example, according to the known methods such as chlorination of hydroxypyrimidines using phosphorus oxychloride (see *Tetrahedron*, Vol. 35, p. 2087, 1979; or *Journal of Heterocyclic Chemistry*, Vol. 20, p. 219, 1983).

[0049] The compounds represented by Formula (III) can be produced, for example, using ketones, sodium cyanide, and ammonium chloride, according to the Strecker Method, which has been disclosed in *Organic Syntheses*, Vol. 3, p. 88, 1955; *Journal of Medicinal Chemistry*, Vol. 9, p. 911, 1966; or *Tetrahedron Letters*, Vol. 17, p. 1455, 1977.

Preparation Process 2

[0050]



(wherein R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , Q, and Z have the same meanings as defined above).

[0051] The compounds represented by Formula (I-1) according to the present invention can be prepared by the reaction of pyrimidine derivatives represented by Formula (VII) with alkanamide derivatives represented by Formula (VIII) in the presence of a base. The present reaction can be carried out in a solvent: this solvent can be the same solvent as described in Preparation Process 1, that does not hinder the reaction.

[0052] The base can be any type of base generally used in this type of reaction. For example, there can be mentioned inorganic bases such as sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, sodium bicarbonate, sodium hydride, potassium hydride, and the like, or organic bases such as triethylamine, trimethylamine, N,N-dimethylaniline, pyridine, and the like.

[0053] The present reaction is carried out at a temperature of -50°C to 150°C, and preferably 0°C to 60°C. The reaction time is preferably in the range from 1 to 30 hours.

35 [0054] In the present reaction, the compounds represented by Formula (VII) can be produced, for example, according to the reaction of pyrimidine derivatives represented by Formula (IV) with thioureas.

[0055] In addition, the compounds represented by Formula (VIII) can be produced, for example, according to the reaction of halogenated alkanoyl halides with amine derivatives represented by Formula (III).

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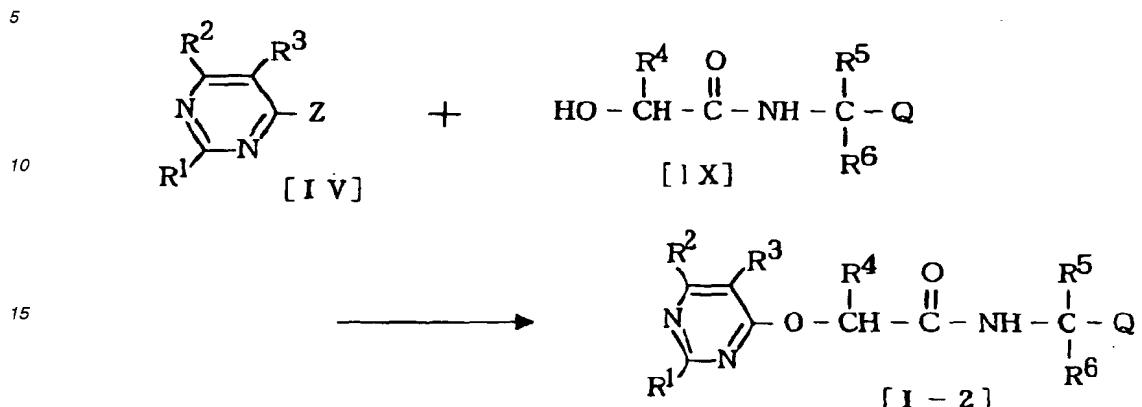
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Preparation Process 3

[0056]



dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 2.6 g of the desired product as an oil.

Reference Example 1-b

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Synthesis of 2-(5-chloro-6-ethylpyrimidin-4-yloxy)acetic acid (Intermediate Compound No. 2)

[0064] Ethyl 2-(5-chloro-6-ethylpyrimidin-4-yloxy)acetate (2.6 g) was dissolved in ethanol (50 ml). A solution containing sodium hydroxide (0.7 g) dissolved in water (10 ml) was added thereto in a dropwise manner, and subsequently the mixture was stirred for an hour at room temperature. After completion of the reaction, water was added to the reaction liquid. Subsequently, the reaction liquid was acidified using citric acid. The organic layer was extracted with ethyl acetate and then was dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The obtained crystals were washed with hexane to give 1.4 g of the desired product having a melting point of 158°C to 159°C.

[0065] Table 16 shows physical properties of pyrimidinyloxyalkanoic acids that are intermediate compounds of the compounds according to the present invention, which are obtained in a manner similar to those described in Reference Example 1-a and Reference Example 1-b.

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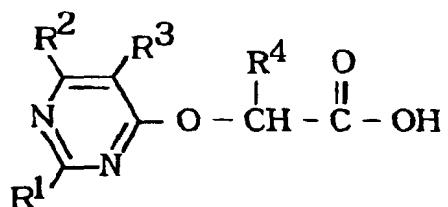
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Table 16



Intermediate Compound No.	R ¹	R ²	R ³	R ⁴	Melting Point (°C)
1	H	Me	Cl	Me	143~145
2	H	Et	Cl	H	158~159
3	H	Et	Cl	Me	126~128
4	H	i-Pr	Cl	Me	140~143
5	H	CF ₃	Cl	H	138~140
6	H	CF ₃	Cl	Me	112~113
7	H	CF ₃	Cl	Et	122~125
8	H	CF ₃	Br	Me	131~134
9	H	CF ₃	Me	Me	113~115
10	Me	CF ₃	Cl	Me	119~122
11	SMe	Me	H	Me	125~127
12	SMe	Et	H	Me	62~ 64
13	SMe	n-Pr	H	Me	56~ 58
14	SMe	i-Pr	Cl	Me	136~139
15	SMe	CF ₃	H	Me	90~ 93
16	SMe	CF ₃	Cl	Me	143~146
17	SMe	CF ₃	Me	Me	112~114
18	Cl	SMe	H	Me	137~139
19	Me	CF ₃	Br	Me	132~135
20	Me	CF ₃	Me	Me	102~104

Preparation Example 2

Synthesis of methyl 2-(1-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)ethylcarbonylamino)-2,3-dimethylbutyrate (Compound No. A-93)

[0066] 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (1.1g) was added to a solution containing 2-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)propionic acid (1.4 g) dissolved in methylene chloride (20 ml), at room temperature, and the mixture was stirred for 10 min. Subsequently, methyl 2-amino-2,3-dimethylbutyrate (0.8 g) was added to the mixture, and the reaction mixture was stirred for 4 hours at room temperature. After completion of the reaction, water was added to the resulting mixture and the methylene chloride layer was washed with water, and subsequently dried over anhydrous magnesium sulfate. The methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 1.3 g of the desired compound having a melting point of 85°C to 86°C.

15 Reference Example 2-a

Synthesis of ethyl 2-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)propionate

[0067] 60% Sodium hydride (0.6 g) was washed with hexane and then suspended in tetrahydrofuran (50 ml). Ethyl lactate (2.0 g) was added to the suspension in a dropwise manner in an ice-cooled bath, and subsequently the mixture was stirred for an hour at room temperature. Subsequently, 4,5-dichloro-6-trifluoromethylpyrimidine (3.1g) was added thereto in a dropwise manner in an ice-cooled bath, and the reaction mixture was stirred for 3 hours at room temperature. After completion of the reaction, water was added to the reaction liquid. The organic layer was extracted with ethyl acetate and then was dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 3.5 g of the desired product as an oil.

Reference Example 2-b

Synthesis of 2-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)propionic acid (Intermediate Compound No. 6)

[0068] Ethyl 2-(5-chloro-6-trifluoromethylpyrimidin-4 yloxy)propionate (3.5 g) was dissolved in 1,4-dioxane (20 ml). A solution containing sodium hydroxide (0.8 g) dissolved in water (10 ml) was added thereto in a dropwise manner in an ice-cooled bath, and subsequently the mixture was stirred for an hour at room temperature. After completion of the reaction, water was added to the reaction liquid. Subsequently, the reaction liquid was acidified using citric acid. The organic layer was extracted with ethyl acetate and then was dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The obtained crystals were washed with hexane to give 2.3 g of the desired product having a melting point of 112°C to 113°C.

Preparation Example 3

40 Synthesis of 2-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)-N-(1-cyanocyclopentyl)propionamide (Compound No. A-80)

[0069] 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (1.1 g) was added to a solution containing 2-(5-chloro-6-trifluoromethylpyrimidin-4-yloxy)propionic acid (1.0 g) dissolved in chloroform (30 ml), at room temperature, and the mixture was stirred for 30 min. Subsequently, 1-amino-cyclopentanecarbonitrile (0.4 g) was added to the mixture, and the reaction mixture was stirred for 10 hours at room temperature. After completion of the reaction, water was added to the resulting mixture and the chloroform layer was washed with water, and subsequently dried over anhydrous magnesium sulfate. The chloroform was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 0.6 g of the desired compound having a melting point of 172°C to 174°C.

Preparation Example 4

55 Synthesis of 2-(5-chloro-2-methyl-6-trifluoromethylpyrimidin-4-yloxy)-N-(1-cyano-1,2,2-trimethylpropyl)propionamide (Compound Nos. A-151 and A-152)

[0070] 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (0.8 g) was added to a solution containing 2-(5-chloro-2-methyl-6-trifluoromethylpyrimidin-4-yloxy)propionic acid (1.0 g) dissolved in tetrahydrofuran (30 ml), at room

temperature, and the mixture was stirred for 30 min. Subsequently, 2-amino-2,3,3-trimethylbutyronitrile (0.5 g) was added to the mixture, and the reaction mixture was stirred for 10 hours at room temperature. After completion of the reaction, water was added to the resulting mixture and the organic layer was extracted with ethyl acetate, and subsequently dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 0.3 g of the A-configurational diastereomer having a melting point of 133°C to 134°C and 0.2 g of the B-configurational diastereomer having a melting point of 137°C to 139°C.

Preparation Example 5

10 Synthesis of 2-(5-chloro-6-isopropylpyrimidin-4-ylthio)-N-(1-cyano-1,2-dimethylpropyl)propionamide (Compound No. B-17)

[0071] 60% Sodium hydride (0.1 g) was washed with hexane and then suspended in dimethylformamide (20 ml). 5-Chloro-6-isopropyl-4-mercaptopurine (0.2 g) was added to the suspension in a dropwise manner, and subsequently the mixture was stirred for an hour at room temperature. Subsequently, N-(1-cyano-1,2-dimethylpropyl)-2-bromopropionamide (0.3 g) was added thereto in a dropwise manner, and the reaction mixture was stirred for 3 hours at room temperature. After completion of the reaction, water was added to the reaction liquid. The organic layer was extracted with ethyl acetate and was then dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 0.3 g of the desired product having a melting point of 73°C to 75°C.

Preparation Example 6

25 Synthesis of 2-(2-allylthio-6-trifluoromethylpyrimidin-4-yloxy)-N-(1-cyano-1,2-dimethylpropyl)propionamide (Compound No. A-251)

[0072] N-(1-cyano-1,2-dimethylpropyl)-2-hydroxypropionamide (1.0 g) was dissolved in tetrahydrofuran (30 ml). 60% Sodium hydride (0.2 g) was added to the solution, and subsequently the mixture was stirred for 30 minutes at room temperature. Subsequently, 2-allylthio-4-chloro-6-trifluoromethylpyrimidine (0.7 g) was added to the mixture, and then the reaction mixture was stirred for 6 hours at room temperature. After completion of the reaction, water was added to the reaction liquid. Subsequently, the organic layer was extracted with ethyl acetate and was then dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 1.1 g of the desired product having a melting point of 96°C to 98°C.

35 Preparation Example 7

Synthesis of 2-(5-chloro-2-methyl-6-trifluoromethylpyrimidin-4-yloxy)-N-(1-isopropyl-1-methyl-2-oxopropyl)propionamide (Compound No. A-334)

40 [0073] 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (0.7 g) was added to a solution containing 2-(5-chloro-2-methyl-6-trifluoromethylpyrimidin-4-yloxy)propionic acid (0.9 g) dissolved in dichloromethane (20 ml), at room temperature, and the mixture was stirred for 10 min. Subsequently, 3-amino-3,4-dimethyl-2-pentanone (0.4 g) was added to the mixture, and the reaction mixture was stirred for 3 hours at room temperature. After completion of the reaction, water was added to the resulting mixture and the dichloromethane layer was washed with water, and subsequently dried over anhydrous magnesium sulfate. The dichloromethane was removed under reduced pressure. The residue was purified by column chromatography on silica gel to give 0.8 g of the desired product having a melting point of 124°C to 126°C.

[0074] The agricultural or horticultural fungicides according to the present invention include pyrimidinylxyloxyalkanamide derivatives represented by Formula (I) as active ingredients. In the case where the compounds according to the present invention are employed as agricultural or horticultural fungicides, the compounds acting as the active ingredients can be formulated appropriately, depending on the purpose. The active ingredient is usually diluted in an inert liquid or a solid carrier, and a surfactant and the like are added thereto, if necessary. The mixture is then formulated in a known manner into, for example, a fine powder, a wettable powder, an emulsifiable concentrate, granules, or the like.

[0075] As the suitable carriers employed in the formulation, there can be mentioned solid carriers such as talc, bentonite, clay, kaolin, diatomaceous earth, white carbon, vermiculite, slaked lime, siliceous sand, ammonium sulfate, urea, or the like; and liquid carriers such as isopropyl alcohol, xylene, cyclohexanone, methylnaphthalene, and the like. As the surfactants and dispersants, there can be mentioned, for example, dinaphthylmethane disulfonate, alcohol sulfates, alkyl aryl sulfonates, lignin sulfonates, polyoxyethylene glycol ethers, polyoxyethylene alkyl aryl ethers, polyoxyethylene

sorbitan monoalkylates, and the like. As auxiliary agents, there can be mentioned carboxymethyl cellulose, and the like. The formulated agricultural or horticultural fungicide according to the present invention can be spread in an appropriately diluted concentration or can be applied directly.

[0076] The agricultural or horticultural fungicides according to the present invention can be employed for spraying 5 stem and leaf portions, application to soil, and submerged application. The proportion of the active ingredient is selected as needed. When formulated into a fine powder or granules, 0.1% by weight to 20% by weight of the active ingredient are preferred. For an emulsifiable concentrate or wettable powder, 5% by weight to 80% by weight of the active ingredient are preferred.

[0077] The rate of application of the agricultural or horticultural fungicide according to the present invention may vary 10 depending on the kind of the compound, the kind of the pest or disease to be controlled, the nature of occurrence of the pest or disease, the degree of damage, environmental conditions, the form of preparation to be used, and the like. When the agricultural or horticultural fungicides of the present invention are applied directly in the form of fine powder or granules, it is recommended that the rate of application of the active ingredient be suitably chosen within the range of 0.1 g to 5 kg per 10 ares, and preferably, in the range of 1 g to 1 kg per 10 ares. In addition, when the fungicides of 15 the present invention are in the form of a liquid such as an emulsifiable concentrate or a wettable powder, it is recommended that the ratio for application of the active ingredient be suitably chosen within the range of 0.1 ppm to 10,000 ppm, and preferably within the range of 1 ppm to 3,000 ppm.

[0078] The agricultural or horticultural fungicide of the present invention can control plant diseases caused by the 20 pathogenic fungi in the *Oomycetes*, *Ascomycetes*, *Deuteromycetes*, and *Basidiomycetes* in the formulation mentioned above. In the following, examples of the fungi will be listed, but are not limited thereto: *Pseudoperonospora* such as downy mildew fungi (*Pseudoperonospora cubensis*), *Sphaerotheca* such as powdery mildew fungi (*Sphaerotheca fuliginea*), *Venturia* such as apple scab fungi (*Venturia inaequalis*), *Pyricularia* such as rice blast fungi (*Pyricularia oryzae*), *Gibberella* such as "Bakanae" disease fungi (*Gibberella fujikuroi*), *Botrytis* such as gray mold fungi (*Botrytis cinerea*), *Alternaria* such as chinese mustard sooty spot fungi (*Alternaria brassicicola*), *Rhizoctonia* such as rice 25 sheath blight fungi (*Rhizoctonia solani*), and *Puccinia* such as rust fungi (*Puccinia recondita*).

[0079] In addition, the compound according to the present invention may be employed alone or in combination with other fungicides, insecticides, herbicides, plant growth modifiers, fertilizers or the like. Next, the representative formulations are illustrated with reference to the following Formulation Examples, wherein all "%" represent "percent by weight".

30 Formulation Example 1: Fine powder

[0080] 2% of Compound No. A-1, 5% of diatomaceous earth, and 93% of clay were uniformly mixed and ground into a fine powder.

35 Formulation Example 2: Wettable powder

[0081] 50% of Compound No. A-7, 45% of diatomaceous earth, 2% of sodium dinaphthylmethane disulfonate, and 3% of sodium ligninsulfonate were uniformly mixed and ground into a wettable powder.

40 Formulation Example 3: Emulsifiable concentrate

[0082] 30% of Compound No. A-12, 20% of cyclohexanone, 11% of polyoxyethylene alkyl aryl ether, 4% of calcium alkylbenzenesulfonate, and 35% of methylnaphthalene were uniformly emulsified, thus yielding an emulsifiable concentrate.

45 Formulation Example 4: Granules

[0083] 5% of Compound No. B-5, 2% of sodium salt of lauryl alcohol sulfate, 5% of sodium ligninsulfonate, 2% of carboxymethyl cellulose, and 86% of clay were mixed and ground. Water was added to the ground mixture, in an amount 50 of 20% based on the total weight of the ground mixture. The resulting mixture was kneaded and formed into granules of 14 mesh to 32 mesh by means of an extrusion granulator, and then dried into the desired granules.

[0084] In the following, the effects exhibited by the agricultural or horticultural fungicides according to the present invention will be explained by reference to Test Examples. In Test Examples, N-(1-cyano-1,2-dimethylpropyl)-2-(pyrimidin-2-yloxy)propionamide disclosed in Japanese Patent Application, First Publication, No. Sho 63 - 132867, was 55 employed as a comparative compound.

Test Example 1

Test on the Preventive Effects for Rice Blast (*Pyricularia oryzae*)

5 [0085] Paddy rice seeds (variety: Aichi Asahi) were sown at a rate of approximately 15 grains each in porcelain pots having a diameter of 7 cm. The seeds were allowed to germinate and grow for 2 to 3 weeks in a greenhouse. A wettable powder prepared according to Formulation Example 2 was diluted with water to a concentration of 500 ppm of the active ingredient, and subsequently the obtained aqueous preparation was sprayed at a rate of 10 ml per pot on the rice seedlings at their 4-leaf stage. After being dried in the air, the seedlings were inoculated by spraying a conidiospore suspension of rice blast fungi (*Pyricularia oryzae*) and immediately placed in a moist chamber at 25°C for 24 hours and subsequently in a greenhouse. On the fifth day after the inoculation, the number of lesions on the fourth leaf was counted. The controlling activity was calculated according to Equation 1. The evaluation results obtained according to the Evaluation Standard shown in Table 17 are shown in Tables 18 to 21.

10

15 Equation 1

[0086]

20

$$\text{Controlling Activity (\%)} = \left(1 - \frac{\text{the Number of Lesions in Treated Plot}}{\text{the number of Lesions in Untreated Plot}} \right) \times 100$$

Table 17

Evaluation	Controlling Activity
A	100%
B	80.0% or more and less than 100%
C	50.0% or more and less than 80.0%
D	less than 50.0%

Table 18

Compound No.	Evaluation	Compound No.	Evaluation
A-1	A	A-66	A
A-5	A	A-68	B
A-6	A	A-69	B
A-7	A	A-70	B
A-12	B	A-71	B
A-13	B	A-72	B
A-15	B	A-73	B
A-16	B	A-74	B
A-18	B	A-75	A
A-21	B	A-76	B
A-22	B	A-77	B
A-25	A	A-78	A
A-26	A	A-79	B
A-27	A	A-80	B

Table 18 (continued)

	Compound No.	Evaluation	Compound No.	Evaluation
5	A-28	B	A-81	B
	A-31	B	A-82	B
	A-32	B	A-83	B
	A-36	A	A-88	B
10	A-38	A	A-89	B
	A-39	B	A-90	B
	A-40	B	A-91	B
	A-41	B	A-93	A
15	A-42	B	A-94	B
	A-43	B	A-95	A
	A-44	A	A-96	A
	A-49	A	A-98	A
20	A-54	A	A-99	A
	A-55	B	A-100	B
	A-58	B	A-101	A
	A-59	B	A-109	B
25	A-61	B	A-110	B
	A-62	B	A-111	B
	A-63	B	A-112	B
	A-64	A	A-113	B
30	A-65	A	A-114	B

35

Table 19

	Compound No.	Evaluation	Compound No.	Evaluation
40	A-116	B	A-199	B
	A-125	B	A-200	B
	A-128	A	A-201	B
	A-129	A	A-202	B
45	A-132	B	A-203	B
	A-136	B	A-204	B
	A-137	B	A-205	B
	A-141	A	A-208	B
50	A-142	A	A-210	B
	A-143	A	A-214	B
	A-144	B	A-215	B
	A-145	B	A-216	B
55	A-149	A	A-219	A

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Table 19 (continued)

	Compound No.	Evaluation	Compound No.	Evaluation
5	A-150	A	A-220	B
	A-151	A	A-223	B
	A-152	A	A-224	A
	A-153	A	A-225	A
10	A-156	A	A-227	A
	A-157	A	A-228	A
	A-158	B	A-229	A
	A-159	B	A-236	B
15	A-166	A	A-237	B
	A-167	A	A-238	B
	A-168	B	A-239	B
	A-169	B	A-240	B
20	A-172	B	A-241	A
	A-176	B	A-242	A
	A-177	A	A-246	B
	A-183	B	A-247	B
25	A-184	B	A-248	B
	A-192	B	A-249	B
	A-193	B	A-250	B
	A-194	B	A-251	A
30	A-195	B	A-252	B
	A-196	B	A-253	A
35				

Table 20

	Compound No.	Evaluation	Compound No.	Evaluation
40	A-254	B	A-294	A
	A-255	A	A-295	A
	A-256	A	A-296	A
	A-257	B	A-297	A
45	A-258	A	A-298	A
	A-259	A	A-299	A
	A-260	A	A-300	A
	A-261	A	A-301	A
50	A-262	A	A-302	B
	A-263	A	A-303	B
	A-264	B	A-304	B
55				

Table 20 (continued)

	Compound No.	Evaluation	Compound No.	Evaluation
5	A-265	A	A-305	A
	A-266	A	A-306	B
	A-267	B	A-307	B
	A-268	A	A-308	A
10	A-269	A	A-309	A
	A-270	A	A-310	B
	A-274	B	A-311	A
	A-275	B	A-312	B
15	A-276	A	A-313	B
	A-277	A	A-314	B
	A-278	A	A-315	B
	A-279	A	A-316	B
20	A-280	A	A-317	B
	A-281	A	A-318	A
	A-282	A	A-319	B
	A-283	A	A-320	B
25	A-284	A	A-321	A
	A-285	A	A-322	B
	A-286	B	A-323	A
	A-287	A	A-324	A
30	A-288	A	A-325	A
	A-291	A	A-326	A
	A-292	A	A-327	A
	293	A- 293	A-328	A
40				

Table 21

	Compound No.	Evaluation	Compound No.	Evaluation
45	A-330	A	B-1	B
	A-333	A	B-2	B
	A-334	A	B-3	B
	A-335	A	B-5	A
50	A-336	A	B-6	B
	A-337	A	B-7	A
	A-338	B	B-10	A
	A-339	B	B-11	A

Table 21 (continued)

	Compound No.	Evaluation	Compound No.	Evaluation
5	A-340	B	B-14	A
	A-341	B	B-15	B
	A-342	A	B-17	B
	A-343	B	B-18	B
	A-344	B	B-21	B
	A-345	A	B-22	B
10	A-346	B	B-25	A
	A-347	B	B-26	B
	A-348	B	B-28	B
	A-349	B	B-29	B
	A-350	B	Comparative compound	D
	A-354	B		
15	A-355	B		
	A-356	B		
	A-360	A		
	A-361	B		
	A-362	B		

30 Test Example 2

Test on the Submerged Application Effects on Rice Blast (*Pyricularia oryzae*)

35 [0087] Paddy rice seedlings (variety: Aichi Asahi) at their 1.5-leaf stage were transplanted at 4 locations at a rate of 3 seedlings per location each in white porcelain pots having a diameter of 9 cm. The seedlings were allowed to germinate and grow in a greenhouse. When the seedlings grew to the 2.5-leaf stage, wettable powders prepared according to Formulation Example 2 were applied into the irrigation water in the pot so that the concentration of the active ingredient was at 300 g per 10 ares. After 10 days from the treatment, the seedlings were inoculated by spraying a conidiospore suspension of rice blast fungi (*Pyricularia oryzae*) and immediately placed in a moist chamber at 25°C for 24 hours and subsequently in a greenhouse so as to induce an attack of the disease. On the fifth day after the inoculation, the number of lesions on the leaf which was in the most advanced stage of development at the time of the inoculation was counted. The controlling activity was calculated according to Equation 1. The evaluation results obtained according to the Evaluation Standard shown in Table 17 are shown in Tables 22 and 23.

45

Table 22

	Compound No.	Evaluation	Compound No.	Evaluation
50	A-6	A	A-129	B
	A-7	B	A-141	B
	A-18	A	A-144	B
	A-21	A	A-149	B
	A-25	B	A-150	B
	A-31	A	A-151	B
55	A-39	B	A-158	A

Table 22 (continued)

	Compound No.	Evaluation	Compound No.	Evaluation
5	A-41	A	A-166	B
	A-42	B	A-196	B
	A-43	B	A-199	B
10	A-49	B	A-219	B
	A-64	A	A-220	B
	A-65	B	A-223	B
15	A-66	B	A-227	B
	A-68	B	A-241	B
	A-69	B	A-251	A
20	A-70	B	A-254	B
	A-74	B	A-255	B
	A-75	B	A-258	B
25	A-76	B	A-259	A
	A-77	B	A-260	A
	A-78	B	A-261	A
30	A-81	B	A-263	A
	A-82	B	A-264	B
	A-91	A	A-266	B
35	A-93	A	A-267	A
	A-95	B	A-268	B
	A-109	A	A-269	A
40	A-110	B	A-270	A
	A-111	A	A-275	B
	A-112	A	A-276	A
45	A-113	A	A-277	B
	A-114	A	A-278	B
	A-116	A	A-282	A
	A-128	A	A-283	B

Table 23

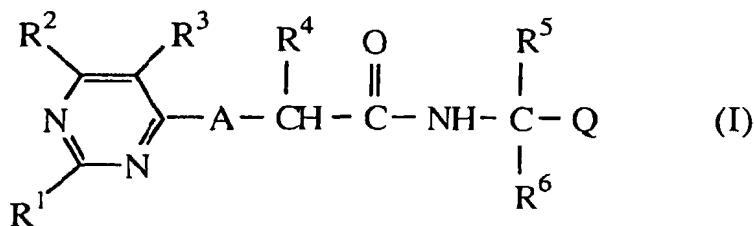
	Compound No.	Evaluation
50	A-284	B
	A-288	A
	A-291	A
55	A-292	A
	A-294	A
	A-295	B

Table 23 (continued)

Compound No.	Evaluation
A-296	A
A-297	A
A-298	A
A-299	A
A-300	A
A-301	A
A-318	A
A-323	A
A-325	B
A-330	A
A-333	A
A-334	A
A-335	A
A-342	B
A-360	B
B-5	B
B-10	B
Comparative compound	D

Claims

1. A pyrimidinyloxyalkanamide derivative represented by Formula (I):



{wherein R¹ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₆ alkylthio group, a C₁ - C₆ alkylsulfinyl group, a C₁ - C₆ alkylsulfonyl group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a C₃ - C₆ cycloalkylthio group, a C₁ - C₆ alkylamino group, a di(C₁ - C₆ alkyl)amino group, a halogen atom, a phenyl group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom), or a phenoxy group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom),

R² represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₂ - C₆ alkenyl group, a C₂ - C₆ alkynyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₂ - C₆ haloalkenyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₄ haloalkoxy group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₄ haloalkoxy group, a C₁ -

C₆ alkylthio group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a C₁ - C₆ alkylamino group, a di(C₁ - C₆ alkyl)amino group, a halogen atom, or a phenyl group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom),

5 R³ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a (C₁ - C₆ alkyl)carbonyl group, a (C₁ - C₆ alkoxy)carbonyl group, a halogen atom, a nitro group, or a cyano group, or

R² and R³, together with the carbon atom to which they are bonded, form a saturated 6-membered ring or an unsaturated 5-membered or 6-membered ring,

10 R⁴ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, or a C₁ - C₄ haloalkyl group,

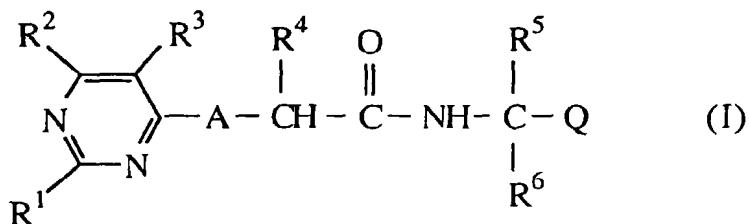
R⁵ and R⁶ represents independently a hydrogen atom, a C₁ - C₆ alkyl group, a C₂ - C₆ alkenyl group, a C₃ - C₆ cycloalkyl group (which may be substituted by a halogen atom or a C₁ - C₆ alkyl group), a C₃ - C₆ cycloalkyl C₁ - C₆ alkyl group, a C₁ - C₆ alkoxy C₁ - C₆ alkyl group, or a C₁ - C₄ haloalkyl group,

15 R⁵ and R⁶, together with the carbon atom to which they are bonded, form a 5-membered ~ 7-membered cycloalkyl group (which may be substituted by a C₁ - C₆ alkyl group), or a heterocycl group (which may be substituted by a C₁ - C₆ alkyl group),

Q represents a cyano group or a group of a formula: -COR⁷ [wherein R⁷ represents a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group (which may be substituted by a halogen atom or a C₁ - C₆ alkyl group), a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, or a C₃ - C₆ cycloalkyloxy group], and

20 A represents an oxygen atom or a sulfur atom.]

2. A pyrimidinyloxyalkanamide derivative represented by Formula (I):



35 {wherein R¹ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₆ alkylthio group, a C₁ - C₆ alkylsulfinyl group, a C₁ - C₆ alkylsulfonyl group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a C₃ - C₆ cycloalkylthio group, a halogen atom, a phenyl group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom), or a phenoxy group (which may be substituted by a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a cyano group, a nitro group, or a halogen atom),

40 R² represents a C₁ - C₆ alkyl group, a C₂ - C₆ alkenyl group, a C₂ - C₆ alkynyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₂ - C₆ haloalkenyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyoxy group, a C₃ - C₆ cycloalkyloxy group, a C₁ - C₄ haloalkoxy group, a C₁ - C₆ alkylthio group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, or a halogen atom,

45 R³ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a halogen atom, or a cyano group, or

50 R² and R³, together with the carbon atom to which they are bonded, form a saturated 6-membered ring or an unsaturated 5-membered or 6-membered ring,

R⁴ represents a hydrogen atom, a C₁ - C₆ alkyl group, or a C₃ - C₆ cycloalkyl group,

55 R⁵ and R⁶ represents independently a hydrogen atom, a C₁ - C₆ alkyl group, a C₂ - C₆ alkenyl group, a C₃ - C₆ cycloalkyl group (which may be substituted by a halogen atom or a C₁ - C₆ alkyl group), a C₁ - C₆ alkoxy C₁ - C₆ alkyl group, or a C₁ - C₄ haloalkyl group,

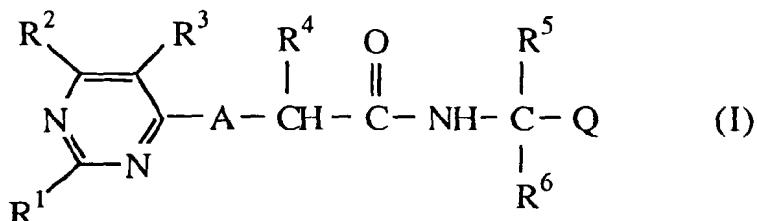
R⁵ and R⁶, together with the carbon atom to which they are bonded, form a 5-membered ~ 7-membered cycloalkyl group (which may be substituted by a C₁ - C₆ alkyl group), or a heterocycl group (which may be substituted by a C₁ - C₆ alkyl group),

Q represents a cyano group or a group of a formula: -COR⁷ (wherein R⁷ represents a C₁ - C₆ alkyl group, a C₃

- C₆ cycloalkyl group (which may be substituted by a halogen atom or a C₁ - C₆ alkyl group), a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₂ - C₆ alkenyloxy group, a C₂ - C₆ alkynyloxy group, or a C₃ - C₆ cycloalkyloxy group), and
 A represents an oxygen atom or a sulfur atom.)

5

3. A pyrimidinyloxyalkanamide derivative represented by Formula (I):



20 (wherein R¹ represents a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₁ - C₆ alkylthio group, a C₂ - C₆ alkenylthio group, a C₂ - C₆ alkynylthio group, a halogen atom, a phenyl group, or a phenoxy group,
 R² represents a C₁ - C₆ alkyl group, a C₁ - C₄ haloalkyl group, a C₁ - C₆ alkoxy group, a C₁ - C₆ alkylthio group, or a halogen atom,
 25 R³ represents a hydrogen atom, a C₁ - C₆ alkyl group, or a halogen atom,
 R⁴ represents a hydrogen atom or a C₁ - C₆ alkyl group,
 R⁵ and R⁶ represents independently a hydrogen atom, a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, or a C₁ - C₄ haloalkyl group,
 R⁵ and R⁶, together with the carbon atom to which they are bonded, form a 5-membered ~ 7-membered
 30 cycloalkyl group (which may be substituted by a C₁ - C₆ alkyl group),
 Q represents a cyano group or a group of a formula: -COR⁷ (wherein R⁷ represents a C₁ - C₆ alkyl group, a C₃ - C₆ cycloalkyl group, or a C₁ - C₆ alkoxy group, and
 A represents an oxygen atom or a sulfur atom.)

35 4. An agricultural or horticultural fungicide including a pyrimidinyloxyalkanamide derivative as recited in Claim 1, 2, or 3 as an active ingredient.

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INTERNATIONAL SEARCH REPORT		International application No. PCT/JP97/03945
A. CLASSIFICATION OF SUBJECT MATTER Int. Cl ⁶ C07D239/34, C07D239/38, A01N43/54		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) Int. Cl ⁶ C07D239/34, C07D239/38, A01N43/54		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CA (STN), REGISTRY (STN), WPIDS (STN)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JP, 62-42974, A (Ube Industries, Ltd.), February 24, 1987 (24. 02. 87), Claims & EP, 212969, A	1 - 4
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed		
Date of the actual completion of the international search January 9, 1998 (09. 01. 98)		Date of mailing of the international search report January 20, 1998 (20. 01. 98)
Name and mailing address of the ISA/ Japanese Patent Office Facsimile No.		Authorized officer Telephone No.